## Piezoelectric characteristics of spray dried PLZT ceramics modified by sodium

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Lead zirconate titanate [Pb (Zr, Ti)O<sub>3</sub>] ceramics modified with lanthanum (PLZT) have been studied by numerous investigators in order to modify the properties of the materials for wide applications in various piezoelectric, electrooptic, ultrasonic, and underwater devices along with actuators, sensors, and smart devices in different fields of science and technology [1–5]. It is well known that the electromechanical properties of PZT depend strongly on additives [6] as well as the morphotropic phase boundary [7]. It has been reported that the electromechanical properties of PZT ceramics could be enhanced by substitution of  $La^{3+}$  on the basis of easy domain wall motion taking place due to the formation of A-site vacancies [8]. In addition, it has also been reported that La<sup>3+</sup> additives influence poling efficiency and consequently the piezoelectric activities of the pure PZT ceramics. (Pb, La) (Zr, Ti)O<sub>3</sub> [PLZT] with 2 at.%  $La^+$  and Zr/Ti ratio 53/47, being in the vicinity of morphotropic phase boundary (MPB), has shown better dielectric and piezoelectric properties. Much work has been reported on PLZT for dielectric and piezoelectric applications but work on the effect of addition of alkali ions Na<sup>+</sup> at A-site of PZT along with  $La^{3+}$  (PLNZT) near MPB has rarely been reported. The present investigation was, therefore, undertaken to study the role of Na<sup>+</sup> ions on the dielectric and piezoelectric properties of PLZT prepared with spray drying technique.

Powders of PLZT and Na<sup>+</sup>-doped PLZT (PLNZT) of the following compositions were prepared by spray drying the composite nitrate solutions of the constituents:

PLZT:  $Pb_{0.97}La_{0.02}[]_{0.01}(Zr_{0.53}Ti_{0.4})O_3$ 

PLNZT:  $Pb_{0.98}(La_{1-x/3}Na_x)_{0.02}(Zr_{0.53}Ti_{0.47})_{0.9950}O_3$ 

where x = 0.0, 0.5, 1.0, 1.5, 2.0, 2.5.

Analytical grade  $Pb(NO_3)_2$ ,  $ZrOCl_2 \cdot 8H_2O$ ,  $TiCl_4$ ,  $NaNO_3$ , and  $La(NO_3)_3 \cdot 6H_2O$  were used as primary raw materials. Lead and lanthanum nitrates were fur-

ther purified by precipitating their aqueous solutions as hydroxide and then dissolving them in dilute HNO<sub>3</sub>. Chlorides of zirconium and titanium (after crystallization in HCl) were converted separately into their nitrates by precipitating their aqueous solutions as hydroxides using NH<sub>4</sub>OH as precipitating agent and redissolving the chloride-free hydroxides in dil. HNO<sub>3</sub>. NaNO<sub>3</sub> was, however, used as such without further purification. Nitrates of Pb, Zr, Ti, La, and Na, taken in appropriate ratio, were spray dried using Eyela Mini Spray dryer, Japan (model SD-1) to get the desired PLZT and PLNZT powders. The spray dried nitrate powders were calcined at 800 °C for 6 hr and pressed into the circular disks of 10-mm diameter and 2-mm thickness at 150 MPa using hydraulic press and 2 wt% PVA binder which was burnt out during sintering at 1100 °C for 3 hr. To prevent PbO loss during high-temperature sintering, an equilibrium PbO vapor pressure was established using PbZrO<sub>3</sub> powder. The sintered disk were lapped and electroded with fired silver paste. The dielectric constant ( $\varepsilon$ ) and tangent loss (tan  $\delta$ ) were measured at different temperatures from 120 to 400 °C at 10 kHz. The disks were polarized using a high dc field of 30 kV/cm for 15 min at 100 °C in a silicon bath. The electromechanical properties of individual pellets were measured using IRE standard [9].

The measured values of dielectric constant ( $\varepsilon$ ), tangent loss (tan  $\delta$ ) (both as a function of variation of temperatures), transition temperature  $T_c$  (°C), diffusivity ( $\gamma$ ), Gaussian diffuseness " $\delta g$ ," electromechanical coupling factor ( $K_p$ ), and piezoelectric strain coefficient ( $d_{33}$ ) at different atomic % of Na<sup>+</sup> additives in PLZT (2/53/47) are listed in Table I.

Addition of Na<sup>+</sup> in PLZT causes significant changes on electrical and electromechanical properties (Table I). The dielectric constant ( $\varepsilon$ ), tan  $\delta$ , Gaussion diffuseness ( $\delta_g$ ) increase with the addition of Na<sup>+</sup> atomic % and show a peak at 1.5 at.% of Na<sup>+</sup> except  $\gamma$ , which peaks at 1.0 at.% of Na<sup>+</sup> whereas transition temperature ( $T_c$ )

TABLE I Measured values of  $\varepsilon$ , tan  $\delta$ ,  $T_c$  (°C),  $\gamma$ ,  $\delta$  g,  $K_p$ , and  $d_{33}$  at different atomic % of Na<sup>+</sup> additives in PLZT

| Sl. no. | Na <sup>+</sup> (at.%) | $\varepsilon_{\rm max}$ | $\tan \delta$ | $T_{\rm c}$ (°C) | γ    | δg   | K <sub>p</sub> | $d_{33} (\times 10^{-12} \text{ C/N})$ |
|---------|------------------------|-------------------------|---------------|------------------|------|------|----------------|--|
| 1       | 0.0                    | 7450                    | 0.052         | 210              | 1.50 | 64.6 | 0.50           | 290                                    |
| 2       | 0.5                    | 7500                    | 0.046         | 225              | 1.60 | 70.3 | 0.51           | 285                                    |
| 3       | 1.0                    | 7598                    | 0.042         | 240              | 1.75 | 68.3 | 0.53           | 298                                    |
| 4       | 1.5                    | 7690                    | 0.048         | 255              | 1.63 | 81.9 | 0.59           | 312                                    |
| 5       | 2.0                    | 7300                    | 0.037         | 275              | 1.44 | 77   | 0.43           | 276                                    |
| 6       | 2.5                    | 7050                    | 0.034         | 285              | 1.38 | 66.4 | 0.41           | 260                                    |



Figure 1 Variation of  $d_{33}$  and  $K_p$  of PLZT (0% Na) and PLNZT as a function of sintering temperature.



Figure 2 Variation of 'd<sub>33</sub>' of PLZT (0% Na) and PLNZT as a function of poling field.

increases continuously with increase in Na<sup>+</sup> concentrations. Similar results were observed in the case of K-modified PLZT [10]. The electromechanical coupling factor ( $K_p$ ) and piezoelectric strain coefficient ( $d_{33}$ ) also peak at 1.5 at.% of Na<sup>+</sup>. This may be attributed to the fact that lattice vibrations also increase with in-

crease in temperature and the interaction between the increase/decrease of vacancies created by  $La^{3+}/Na^+$  ions doping may be responsible for the change in the  $\varepsilon$ -value and the corresponding  $K_p$  and  $d_{33}$  values. These values going to its maximum for 1.5 at.% Na<sup>+</sup> doping in PLZT may be the result of the fact that predominance

of vacancies (created by  $La^{3+}$  ions) reduces beyond the 1.5 at.% Na<sup>+</sup> addition.

Fig. 1 shows variation of piezoelectric strain coefficient  $d_{33}$  and electromechanical coupling factor  $K_p$  with sintering temperatures of PLZT (0% Na) and PLNZT ceramics. It is obvious (Fig. 1) that the sintering temperature of 1100 °C is the most suitable temperature for the pellets to get the maximum value of  $d_{33}$  and  $K_p$ . This is due to the fact that beyond 1100 °C agglomeration of particles starts taking place with the consequent increase of porosity of the ceramics [1, 11–12].

The variation of piezoelectric stain coefficient with poling (high dc) field for PLZT (0% Na) and PLNZT for various concentrations of Na<sup>+</sup> is shown in Fig. 2. It is observed that the  $d_{33}$  value increases with the increase of applied dc field and it shows peaks at 30 kV/cm for all Na<sup>+</sup> concentrations. This is due to the increase in the mobility of ferroelectric domain walls with increases of dc field and at 30 kV/cm, polarizability comes to a saturation point [13]. Thus, it can be concluded that lead lanthanum zirconate titanate (PLZT) ceramics added to sodium ions may find special applications in piezoelectric, ultrasonic, and underwater transducer devices.

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Received 12 November 2003 and accepted 11 February 2004